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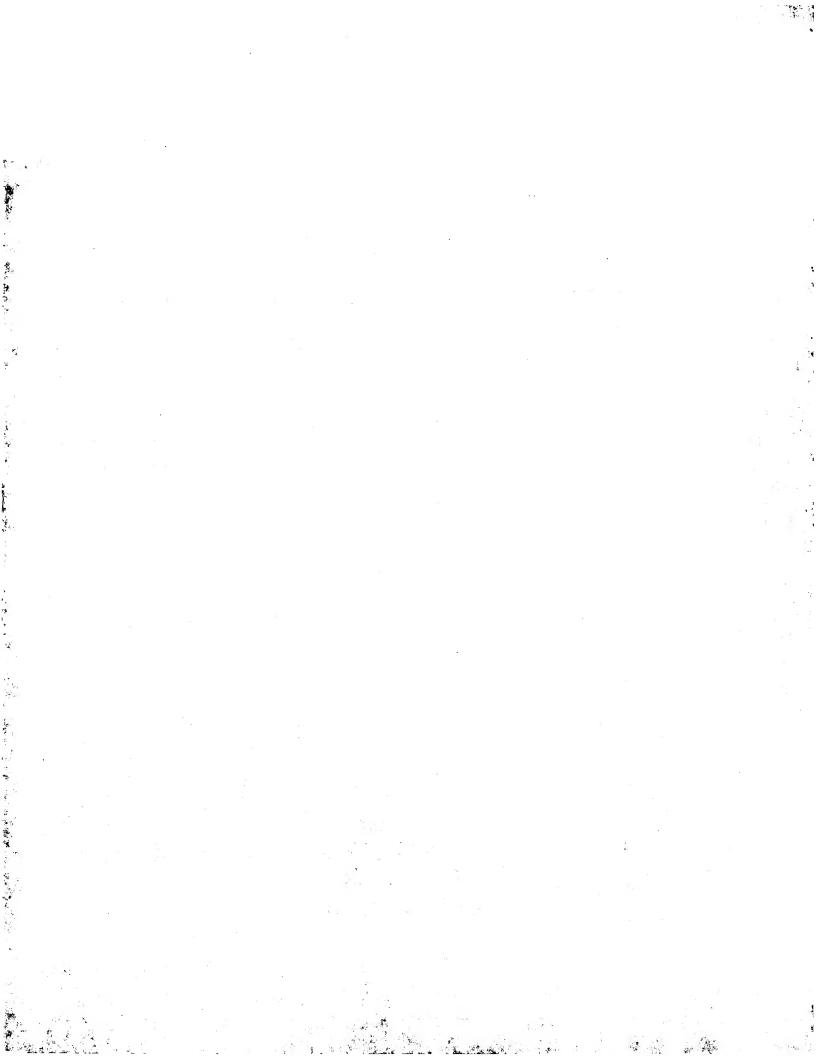
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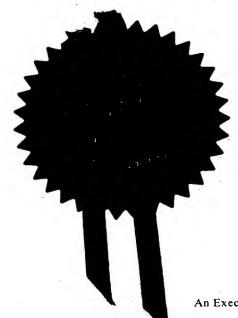
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Patents ADP number (If you know it)

7438275001 United Kingdom

If the applicant is a corporate body, give the country/state of its incorporation

Title of the invention

Particulate Carrier for Biocide Formulations

Name of your agent (if you have one)

"Address for service" in the United Kingdom to which all correspondence should be sent (including the postcode)

Anthony Robert Collingwood

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Particulate carrier for blocide formulations

This invention relates to an inorganic particulate carrier particle for use as a vehicle for introducing blocides into iliquid-based media such as paints, lacquers, plastisols, oil drilling fluids and surface cleaning compositions. The term "blocide" as used herein is to be understood to refer to agents such as germicides, bactericides, fungicides, algicides and the like, which are used for their ability to inhibit growth of and/or destroy blological and/or microbiological species such as bacteria, fungi, algae and the like.

Blocidal agents (blocides) capable of protecting paint, lacquer, plastisol, oil drilling fluids and surface cleaning compositions are well known in the art. US-A-4129448 and 4165318 are illustrative of prior art disclosing the use of blocides to stabilise mildew growth in acrylic emuision polymer paints. US-A-3699231 discloses the use of an aidehyde/carbamate mixture to inhibit bacterial growth. Other inhibiting admixtures are known containing isothlazolones and chlorinated derivatives of which US-A-3929561 and US-A-4295932 are examples. All the above disclosures describe a method for protecting the bulk formulation by adding the blocide directly to the composition.

Attempts to control the release of blocide to inhibit bacterial and fungal growth have centred around the use of sol get chemistry to entrap the blocide but allow release thereof by diffusion from the hydrogel network. This approach is exemplified by EP-A-0602810, EP-A-0738249, GB-A-2235462 and GB-A-1590573 and US-A-5229124. Applications of sol get entrapment technology for the controlled release of blocide have however been limited owing to the need for the components, that form the encapsulation system "in situ", to be included in a particular formulation and be compatible with the remainder of the ingredients.

Another approach involving the encapsulation of organic liquids such as perfumes, food flavours, pesticides and fungicides is disclosed in US-A-4579779 in which the organic liquid is combined with particles of amorphous silica having a pore size distribution wherein 50% of the integrated micropore volume is constituted by micropores having a radius up to 500 Angstrom Units (AU), the liquid and particles being combined in such as way that droplets of the organic liquid are encompassed within a shell of silica particles. The silicas employed are Tokusii PR and Tokusii NR, made by Tokuyama Soda Co. Ltd. The average particle size and BET surface area for Tokusii PR and Tokusii NR are 100 and 130 microns respectively and 198 and 195 m²/g respectively.

The impregnation of mineral particles with blocides is also known. US-A-4552591 describes a composition intended to protect polymer dispersions used in oil field water treatment. This composition comprises a liquid blocide adsorbed on mineral adsorbents, granular or bead-like in nature, such as distomaceous earth, silica, metal oxides (alumina bauxite, magnesia iron oxide), clays, zeolites, resins and waxes. Apart from a general reference to "well known adsorbents having a high degree of surface area", no mention is made of key properties such as surface area, pore volume, pore size, pore size distribution. The preferred adsorbent is distomaceous earth and whilst this material has a high propensity

for liquids, there is no evidence that the carrier particle will retain blocide within its pore system and provide controlled release to an aqueous based composition.

The present invention seeks to provide an improved blocide-carrying carrier particle.

According to one aspect of the present invention there is provided a particulate composition of matter comprising porous inorganic carrier particles having blocide adsorbed within the pore system thereof and having a retention factor (as defined herein) of at least 0.6, preferably at least 0.8.

The retention factor, R, referred to above is determined from the equation R = A/P, where A represents the percentage active ingredient by weight remaining in the pore system after contacting with water according to the conditions defined herein and P represents the potency (Minimum Inhibition Concentration in mg of active ingredient per litre) of the blocide determined with respect to the reference microorganism Aureobasidium pullulans using the procedure defined herein.

The usefulness of the inorganic carrier particle will depend on the particular blocide being used, its effectiveness at various activity levels and the quantity of blocide (active ingredient) adsorbed and retained in the pore system. The amount of blocide deemed to be effective in the pore system will depend on the potency of the blocide, that is, the minimum concentration of active ingredient to prevent microbial or fungal growth. For the purposes of the present invention, the reference microorganism is Aureobasidium pullulans. Commonly used blocides are 2-Octyl-4-isothiazolin-3-one (OIT) and a blend of 2-Methyl-4-isothiazolin-3-one (MiT) and 5-Chioro-2-Methyl-4-isothiazolin-3-one (CIT), known as (CIT/MIT). For these blocides, the Minimum Inhibition Concentration (MIC) is 36 and 5 mg of active ingredient per litre for OIT and CIT/MIT respectively.

Prior to contact with liquid media into which the particles are to be introduced, the particles preferably carry at least 30% by weight of blocke in aqueous solution or water/organic solution.

Such particles will usually be chemically inert with respect to the liquid media into which they are introduced.

A feature of the invention is the ability of the particles, when formulated into solvent or aqueous based compositions, to retain the blocide within the pore system thereof to such an extent that release of the blocide into the tiquid media is sufficiently retarded in order to provide an extended period of blocidal, e.g. bactericidal and/or fungicidal, activity.

Preferably the inorganic particles have an activated micropore system. Under the IUPC system, a micropore is one having a diameter of no more than 30 AU, activation usually being achieved by thermal treatment. Whilst not wishing to be bound by theory it is thought that the carriers that contain an activated micropore system are capable of adsorbing the blockde molecules in preference to water and other substrate molecules.

In order to secure appropriate retention of biocide, a preferred inorganic carrier particle has a pore area of at least 25 m²/g, preferably at least 30 m²/g, more preferably at least 40 m²/g, and up to about 300 m²/g, e.g. 50 m²/g to 250 m²/g, in the pore size range of from about 20 to about 60 Angstroms,

and a BET surface area of at least 200 m 2 /g and more preferably at least 300 m 2 /g, typically 350 m 2 /g to 1200 m 2 /g.

The porous inorganic carrier particles preferably have a blocide adsorption capacity of at least 10%, more preferably at least 15% and most preferably at least 20% by weight of the carrier particle plus blocide. The adsorption capacity in this instance is the amount of blocide which is retained in the pore system of the porous inorganic carrier particle when the particle containing the blocide is contacted with water as defined herein. The desired blocide adsorption capacity of the carrier particle will, in practice, depend on the particular blocide employed and its potency.

The biocide generally comprises a composition to control and prevent the germination and growth of bacteria, fungi and algae and include the following suitable chemical types: aldehydes. formaldehyde condensates, triazines, phenolics, carbonic acid esters, amides, eg, N-(3,4-dichior phenyl)-N,N-dimethyl urea, carbamates, eg, methyl-N-benzimidazol-2-methylicarbamate, thiocarbamates thiocyanates, dibenzamidines, pyridine derivatives, triazoles, thiazoles, isothiazolones, eg, 2-methyl-4-isothiazolin-3-one, N-haloalkylthio compounds, eg, N-dichiorofluoromethylthiophthalimide and the like. The isothiazolin-3-ones are the presently preferred blocides.

Suitable isothiazolin-3-ones include 2-methyl-4-isothiazolin-3-one, 2-ethyl-4-isothiazolin-3-one, 2-propyl-4-isothiazolin-3-on, 2-butyl-4-isothiazolin-3-one, 2-amyl-4-isothiazolin-3-one, 5-chloro-2-methyl-4-isothiazolin-3-one, 5-iodo-2-methyl-4-isothiazolin-3-one, 5-chloro-2-butyl-4-isothiazolin-3-one, 5-bromo-2-ethyl-4-isothiazoline-3-one, 5-iodo-2-amyl-4-isothiazolin-3-one, 1,2-benzisothiazolin-3-one, 2-n-octyl-4-isothiazolin-3-one, 4,5-dichioro-2-n-octyl-4-isothiazolin-3-one and other similar enalogues and homologues within the genus.

Advantageously the blocide is selected from a mixture of 5-chloro-2-methyl-4-isothiazolin-3-one and 2-methyl-4-isothiazoline-3-one, typically in the weight ratio of between 1.5 and 2.5:1, e.g. in the range 2.7 to 3:1; 2-n-octyl-4-isothiazolin-3-one, or 4,5-dichloro-2-(n-octyl)-4-isothiazolin-3-one.

The carrier particles suitable may be constituted by amorphous silicas, amorphous aluminas, pseudoboehmites (a form of microcrystalline aluminium hydroxide), Y-zeolites or dealuminated Y-zeolites. In the case of the latter, the Si:Al ratio should be in the range from preferably from about 5:1 to about 33:1.

Adsorption of the biocide is usually achieved by mixing the carrier particles with the biocide and such mixing may be carried out in a variety of ways known to those skilled in the art, for example spraying the biocide solution onto the porous inorganic particles in a rotary drum, or while conveyed on a conveyor belt. Non limiting examples of powder mixers include, Nauter conical mixers, double cone mixers, trough mixers, fluid bed mixers and various rotating blade vessel mixers. In all these mixers the powder charge is fluidised by a paddle, screw, air agitation or by mechanical rotation. The blocide solution is sprayed onto the particles and mixing continued until the take-up of blocide solution to the desired level is obtained (usually so that the porous inorganic particles maintain a free flowing

consistency). The blockde/inorganic particle composition can then be dropped by gravity into suitable containers.

The amount of blocide added to the porous inorganic carrier particles will depend on the particular blocide being used and its effectiveness at various activity levels. Thus the concentrated blocide is usually diluted to afford an activity level commensurate with microbial inhibition. Further, a solution 10% by weight blocide in an appropriate solvent (10% active) has blocidal properties which will vary depending on the particular blocide and the amount of the solution which is added to a particular microbial culture. Accordingly, a blocide may be added to the porous inorganic particles undiluted (100% active) or it may be diluted with a solvent to a lower activity (as low as 10% active). This is particularly the case for the isothiazolin-3-ones where the pure blocide is usually diluted with water or water alcohol blends to achieve levels of from about 1 to about 60%, preferably from about 2 to about 50%, e.g. about 5 to about 25%, active by weight.

The blocide-impregnated carrier particles are particularly useful in surface coating and sealant compositions, e.g. paints, lacquers, sealants and plastisols.

According to a second aspect of the invention there is provided a surface coating or sealant composition incorporating a biocide-carrying particles in accordance with said first aspect of the invention.

Typically the composition comprises a film forming material, which is usually polymeric in nature, a solvent and various additives, such as pigments, dyes, dispersing agents, surfactants and antifoaming agents. The total non-volatile content of the composition, usually referred to as the solids content, can vary widely. Often it is desirable that the non-volatile content be at least 30% by weight of the formulation in order that a practical amount of film forming material is present in the dried/cured film. The compositions often can be satisfactorily formulated at a non-volatile content as low as 10% by weight or as great as 100% by weight but, in the latter case, thinning may be necessary at this concentration to provide satisfactory application. The preferred non-volatile concentration is from about 30 to about 65% by weight, irrespective of whether the solvent is organic-based or water.

Where the composition is water based, it generally comprises an aqueous dispersion of the polymer in an emulsion form as the major film forming component whereas a solvent borne composition usually employs the polymer per se, e.g. in solution. Examples of such polymers include homopolymers and copolymers of: (1) vinyl esters of an aliphatic acid having 1 to 18 carbon atoms, especially vinyl acetate (2) acrylic acid esters and methacrylic acid esters of an alcohol having 1 to 18 carbon atoms, especially methyl acrylate, ethyl acrylate, butyl acrylate, 2-ethyhexyl acrylate, methyl methacrylate, ethyl methacrylate; and (3) mono and di-ethylenically unsaturated hydrocarbons, such as ethylene iso-butylene, styrene and aliphatic dunes, such as butadiene, isoprene and chloroprene.

Poly(vinyl acetate) and copolymers of vinyl acetate with one or more of the following compositions: vinyl versatate or vinyl esters of fatty acids having 3 to 18 carbon atoms, vinyl chloride, vinylidene chloride, styrene, vinyl toluene., acrylonitrile, methacrylonitrile, mono or di-fumeric or -maleic acid esters, such as of the alkanols having 1 to 4 carbon atoms, including for example, monomethyl

fumerate, diethyl maleate or fumerate, dibutyl maleate or monobutyl maleate, or one or two of the acrylic and methacrylic esters mentioned above may be used as the film forming components where the paint is aqueous based. Similarly, copolymers of one or more of the acrylic or methacrylic acid esters mentioned above with one or more of the following monomers: vinyl acetate, vinyl esters of higher fatty acids, the mono or di-alkyl esters of itaconic acids, the mono or di-alkyl esters of fumeric acid or the mono or di-alkyl esters of maleic acid, such as esters of methanol ethanol, or butanol, vinyl chloride, vinylidene chloride, styrene, vinyltoluene, acrylonitrile and methacrylic nitrile may be used in the composition according to said second aspect of the invention.

The surface coating or sealant composition can contain additional materials to vary the properties and to adapt the composition for different uses. For example plasticisers can be added together with a range of pigments and dyes. The relative proportions of vehicle to pigment may fall in a wide range, such as from a ratio of 1:20 to 20:1 but for most formulations the range is 1:5 to 5:1. Cellulose derivatives such as methyl cellulose, carboxylmethyl cellulose or hydroxyethyl cellulose can be used as bodying agents. Antifoam agents can be included to control foam generation caused by the presence of surfactants to assist the dispersion of pigments and dyes. Other suxiliary materials that may be used include dispersing agents, such as aromatic sulphonates condensed with formaldehyde, humectants such as water soluble gums, glycol laurate, propylene glycol, diethylene glycol and the like, thickeners/body agents, perfume and like materials including neutralising and masking agents which are used to overcome odours or impart pleasant odours; other resinous materials such as drying oils or latices of styrene or of styrene and butadiene.

The amount of porous inorganic carrier containing the blocide incorporated in a surface coating or sealant composition will vary according to factors such as the composition itself, particular inhibitor composition, the conditions of use of solvent, water or polymer dispersion and the extent of prior contamination with micro-organisms, the time period of growth inhibition desired, the requirements of the Health and Safety exposure limits. Usually, to afford adequate protection for many applications, the amount of blocide-containing carrier particle added to the surface coating or sealant composition will be such that the active blocide constitutes from about 0.01% to about 3%, e.g. 0.01% to 2%, by weight of the composition.

The porous inorganic carrier containing the blocked may be incorporated into the surface coating or sealant composition by adding the particulate carrier (i) to water used in the formulation, (ii) to the polymer dispersion or (iii) to the total formulation. This is optimally carried out in a containing vessel which can be readily agitated with a high speed disperser such as a Silverson mixer.

The blocide-containing particles of the invention also have application in surface cleaning compositions in order to enhance the performance of such compositions. Accordingly, in a third aspect of the present invention there is provided a surface cleaning composition incorporating blocide-carrying particles according to said one aspect of the invention.

The surface cleaning composition preferably comprises an aqueous dispersion of a surfactant and an inorganic builder such as an aluminium silicate or zeolite and, optionally, other components such as one or more of the following: water soluble complex formers or precipitating agents for calcium ions; abresives; water soluble or water dispersible organic solvents; hydrotropes; and soil suspending agents.

Suitable surfactants of the sulphonate type are alkly benzene sulphonates in which the alkyl group has from 9 to 15 carbon atoms, alkane sulphonates, esters of alpha-sulpho fatty acids, sulphuric acid mono esters of primary allphatic C_{10} to C_{12} alcohols, sulphated fatty acid alkanol amides, fatty acid mono glycerides with C_{10} to C_{20} fatty acids and sulphates of primary or secondary allphatic C_{10} to C_{20} alcohis reacted with 1 to 6 moles of ethylene oxide. Surfactants having anionic groups may be present in the form of their sodium, potassium, and ammonium salts or in the form of water soluble salts of organic bases, such as mono -, di or tri ethanol-amine.

Suitable non-ionic surfactants are addition products of ethylene oxide and an alphatic C_{10} to C_{20} alcohol or an alkyl phenol, fatty amine or fatty acid, ethoxylated products of allphatic alcohols, C_{10} to C_{20} oxoalcohols and secondary allphatic alcohols having 12 to 18 carbon atoms. Suitable non-ionic surfactants also include surface active amine oxides such as N-dodecyl-N, N-dimethyl amine oxide, N-tetradecyl-N, N-dihydroxy ethyl amine oxide, N-hexadecyl-N, N-bis (2,3 - dihydroxy-propyl) amine oxide.

In addition to a hydrophobic, generally aliphatic group, suitable zwitterionic surfactants that not only contain hydrophilic acidic groups, but also basic groups are useful. Zwitterionic compounds having four-fold substituted ie: quaternary ammonium group belonging to betaine group can also be formulated into the composition. Particularly useful are the carboxy, sulphonate and sulphate betaines of nitrogen. Typically representative examples of zwitterionic surfactants are compounds of 3-(N-hexadecyi-N, N-dimethylammonio)-propane sulphonate, 3-(N-coconut-alkyi-N, N-bis-(2,3-dihydroxy propyi)-ammonio)-propane sulphonate.

Sultable complex formers or precipitating agents for calcium ions or heavy metal ions include inorganic agents, such as pyrophosphate, tripolyphosphate, higher polyphosphates and metaphosphates, and the organic agents, such as salts of aminopoly-carboxylic acids, for example, nitrile tri acetic acid, ethylene-diamine-tetra-acetic acid, of citric acid, gluconic acid; of carboxy-methyl-ether-carboxylic acids having molecular weights in excess of 350, for example, poly acrylic acid, poly-alphahydroxyacrylic acid. Also useful are the water soluble salts of the phosphono-alkane-polycarbxylic acids and the amino-and hydroxy-substituted alkane polyphosphonic acids.

Soil suspending agents that may be employed are generally water soluble colloids, such as water soluble salts of polymeric carboxylic acids, glue, gelatine, salts of either-carboxylic acids or ether sulphortic acids of starch and cellulose, or salts of acidic sulphuric acid esters of cellulose or starch. Polyamides containing water soluble acidic groups are also sultable for this prupose. In addition, soluble starch preparations and starch products, such as decomposed starches aldehyde starches and polyvinyl pyrrolldon may also be used.

The most suitable organic solvents that may be employed are alcohols and ether alcohols which are water-soluble or can be emulaified with water, for example ethanol, isopropyl alcohol, butanol, amyl alcohol, ethylene glycol, diethylene glycol.

The amount of porous inorganic carrier containing the blocide incorporated in the surface cleaning composition will vary according to factors such as those mentioned previously in connection with surface coating compositions. Typically, a surface cleaning composition according to said third espect of the invention will incorporate an amount of blocide-carrying particles such that the blocide constitutes about 0.1 to about 3% by weight of the cleaning composition.

The blocide-containing particles may be incorported into the surface cleaning composition by adding the particulate carrier (i) to the water used in the formulation, (ii) to the additive dispersion, or (iii) to the total formulation. The addition of the inorganic carrier particulate containing the blocide is optimally carried out in a containing vessel which can be readily agitated with a rotating blade, propellor or turbine.

Definitions and Procedures

The porous inorganic particulate carrier blocide compositions of the invention are defined in terms of the properties and texture of the porous inorganic particulate together with their capability to adsorb blocide and retain it within the specifically selected pore size range.

I) Weight Mean Particle Size

The weight mean particle size of the porous inorganic carrier particulate is determined using a Maivern Mastersizer model X, with a 45mm lens and MS15 sample presentation unit. This instrument, made by Maivern Instruments, Maivern, Worcestershire, United Kingdom uses the principle of Mie scattering, utilising a low power He/Ne laser. Before measurement the sample is dispersed ultrasonically in water for 5 minutes to form an aqueous suspension. This suspension is stirred before it is subjected to the measurement procedure outlined in the instruction manual for the instrument, utilising a 45 mm lens in the detector system.

The Malvern Mastersizer measures the weight particle size distribution of the silica or reference material. The weight mean particle size (d_{50}) or 50 percentile, the 10 percentile (d_{10}) and the 90 percentile (d_{90}) are readily obtained from the data generated by the instrument.

II) BET surface area

Surface area is determined using standard nitrogen adsorption methods of Brunauer, Emmett and Teller (BET), using a single point method with a Sorpty 1750 apparatus supplied by Carlo Erba company of Italy. The sample was outgassed under vacuum at 270°C for 1 hour before measurement.

III) High Performance Liquid Chromatography (HPLC)

High Performance Liquid Chromatography(HPLC) was used to evaluate the concentration of blocide in a solvent system. Typically the sample of blocide in a liquid sample is loaded onto a Nucleosil 10C₁₈ column and eluted along the column at a fixed flow rate by the use of eluting solvents and a pump. As with other chromatographic methods the materials loaded onto the column will pass

through the column packing at different rates. The time at which a material exits the column is known as the retention time and is characteristic of the compound being analysed and the method being used. As the components of the mixture exit the column they are analysed by an accurate internal or external UV/Visible spectrophotometer.

The use of HPLC involves three separate steps. Firstly the mobile phase/diluent has to be selected and then prepared. Choosing the most appropriate solvents is often a matter of trial and error. For OIT, methanol, water and acetic acid in a 65:35:0.2 ratio has been found to be the most suitable, whilst for CIT/MIT, methanol, water and acetic acid in a 65:35:0.4 ratio gives the best result. Preparation of the mobile phase involves adding the the relevant quantities of solvent, ensuring complete mixing and then degassing using an ultrasonically agitated bath. The second step is calibration of the HPLC equipment which is achieved by analysing a sample of known composition and concentration. The analytical standard should be prepared using the same mobile phase/diluent as that to be used in the determination. The final step is to produce calibration graphs for the biocides used in the study. This is achieved by preparing samples of known quantities of biocide and obtaining concentration values from the generated chromatograms. These concentration values are then used to plot a calibration graph for each biocide used in the study. Test samples can then be run on the HPLC and the quantity of biocide present in the test solutions derived. This method was used to validate the UV/VIS spectroscopic method referred to below.

iv) UV/VIS Spectroscopy

Second derivative UV-Visible spectroscopy has been employed as an alternative to HPLC for determining the blocide concentration in either water or propylene glycol/water. The advantages of this technique being its simplicity of use and high degree of accuracy. Also single- and multi-component systems can be analysed. The deflections in second derivative spectrum are not proportional to the absorption values in the original spectrum. Rather they are proportional to the slope of the latter providing it was scanned in absorption mode. Positive and negative slopes are shown as positive and negative deflections in the derivatised spectrum. It is the position of the absorption and the relation of the extremes which are of interest in this method. Its application here was as a means of monitoring the amount of biocide leached into a solvent and as such the procedure used to evaluate the results was to measure the amplitude of deflection (delta A) on a peak to peak basis. In this method, the absolute distance between a maximum and an adjoining minimum is determined as a characteristic of the species under investigation. This distance is then compared with a standard calibration to attain the concentration of the sample. A Perkin-Elmer lamda 7 spectrophotometer was used in conjunction with the lamda 16 WINDOWS software package to determine accurate values of (delta A) from the derivative spectra obtained. The software package can be obtained from Perkin Elmer of Post Office Lane, Beaconsfield, Bucks HP9 1QA, United Kingdom.

v) Pore area in a pore size rance

Nitrogen adsorption isotherm is determined using a multi-point method with ASAP 2400

apparatus supplied by Micrometrics of the USA. The samples are outgassed under vacuum at 270 °C for at least one hour before measurement. This apparatus also enables the pore size distribution from the adsorption branch of the isotherm to be calculated. This can be expessed in terms of the cumulative pore area contained in a given range of pore size. The pore area within the pore size range 20 to 50 Angstoms can be readily derived.

vi) Leaching

In this work the biocide carrier particles were added to water in the ratio of 0.3:1 blocide carrier to water. An amount of this blend was chosen to ensure that when it was added to 1000 ml of water the solubility maximum for OIT of 400 ppm would not be exceeded. CIT and MIT are both more water soluble than OIT.

The method employed was as follows: 0.75 g of blockde was added to 2.6 g of carrier material. This was then transferred to a vessel and homogenised by rotating the vessel on rollers for 8 hours. The homogenised blend was then charged into 1000 ml of distilled water, stirring continuously. Aliquots of the slurry were taken at intervals of 0.5, 1, 2, 4, 6, 12, 30, 60 and 90 minutes and it was established that after 60 minutes the elution curve plateaued and to ensure that equilibrium had been secured, the experimental work was conducted using a 90 minute equilibrium time. These were filtered and diluted to a concentration within the range of the calibration curve for the respective blockde under investigation. The samples were then analysed by 2nd derivative UV/VIS spectroscopy as described above.

vii) Fungicidal Assessment Of Coatings

This method was developed so that fungicidal activity of a blocide within a coating could be evaluated. It can be used to investigate the effects of, for example, water leaching, film weight and concentration on the diffusivity of the blocide.

Coatings with and without the blocide were brush applied in two separate coatings (24 hours between each coating) to one side of rubber discs discs (3.7 cm in diameter). Once dry, the discs were immersed in 30ml of sterile distilled water for 1 hour and then dried in a laminar flow cabinet overnight. Potato dextrose agar plates in Petri dishes were prepared, dried and separately inoculated with 1ml of spore suspensions containing as 10⁶ cfu/ml of each test fungus. Four fungal species were included in the study: Aureobasidium pullulans (FS103), Rhodotural robra (FS83), Cladosporium cladosporioldes (IMI71749R) and Atternaria atternata (IMI78517). The suspensions were allowed to adsorb into the agar to remove surface wetness. The coated discs were then placed, with coated face down, at the centre of the agar carrying Petri dish. The assembly of plate/disc was then stored in refrigerator for 24 hours to allow diffusion of the blocide to occur. The plates were subsequently placed in an incubator at 26 °C for 5 days. Zones of inhibition were calculated by summing the radial distances as measured from the edge of the filter paper to the fungal growth front at diametrically opposite locations of the disc.

viiii) Minimum Inhibition Concentration

The minimum inhibition concentration (MiC) is a measure of the potency of a blocide and comprises the minimum concentration (ppm) of a blocide active or formulation (normally quoted in terms of the active ingredient) determined experimentally *in vitro* to prevent the growth of a pure culture of a reference microorganism. Although the experimental design may differ, as there are no standard methods, the general principles of the necessary procedures remain the same.

A standardised culture of the microbe under test is prepared. Aliquots are added to a suitable liquid nutrient medium in stoppered glass tubes containing a range of concentrations of the blocide active. The mixtures are incubated at a standard controlled temperature to encourage growth of the microbes for a standard time, usually between 24 to 48 hours. The tubes are removed and assessed for growth/no growth by measuring the increase in the optical density of the medium caused by an increase in the number of microoganisms present. Other detection methods such as total viable counts or visual examination may be used. The concentration at which no growth is detected is the minimum inhibition concentration (MIC) for that particlar blocide and reference organism. The measurements can be refined by repeating the test using intermediate concentrations of the biocide, between those already chosen as being representive of the killing range, in order to more accurately define the MIC.

Specific Description Of The Invention

The following examples serve to illustrate but not limit the present invention. In all the examples, unless otherwise stated the following blocides have been used. The definition of isothiazolone blocide as used here corresponds to the general structural formula shown below.

$$R_2$$
 $N-R_1$

4-R₂-5-R₃-2-R₁-4-Isothiazolin-3-one

 R_1 is an alkyl group with the formula $-(CH_2)_nCH_3$ where n = 1 to 8

R₂ can be either H or Halogen, and

R₃ can be either H or Helogen

Some examples of this class of compounds are included below

DCOIT

4,5-Dichloro-2-Octyl-4-Isothiazolin-3-one (DCOIT)

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ACTICIDE® 45 - OIT

2-Octyl-4-Isothiazolin-3-one (OIT)

46.9 % OIT in Propylene Glycol

ACTICIDE® 14L - CIT/MIT

2-Methyl-4-Isothiazolin-3-one (MIT)

5-Chloro-2-Methyl-4-Isothiazolin-3-one (CIT)

10.3 % CIT / 3.8% MIT (14.1 % Active), in Water

ACTICIDE® TL666 - CIT/MIT

2-Methyl-4-Isothiazolin-3-one (MIT)

5-Chloro-2-Methyl-4-Isothiazolin-3-one (CIT)

2.5 % or less Active in glycol as a ratio of 3:1 CIT/MIT

Example 1

To allow acreening tests to be conducted on a small laboratory scale the blocide impregnated inorganic particulate carriers hereinafter described were made by adding the appropriate amount of blocide dropwise, 0.7g of blocide to 2.5g of carrier material, and then homogenised by rotating the vessel containing the blend on rollers for 8 hours. The impregnated inorganic particulate carrier particles were then charged into 1000ml of distilled water and allowed to equilibriate for 90 minutes

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under stirring. The suspension was filtered and the level of blocide analysed in the filtrate using 2nd derivative UV/VIS spectroscopy as previously described. By simple difference the % retained by the carrier could be calculated. The results obtained for the carrier materials impregnated with OIT are given in Table I.

Table I

| Sample No. | Material | SI:Al Retio | B.E.T. Surface Area m²/g | Average Particle Size microns | OIT Retention % By Weight |
|------------|-------------------------|-------------|-----------------------------|-------------------------------------|------------------------------|
| SD1866 | Amorphous Silica | | 728 | 4.6 | 14 |
| SD1866(c) | Amorphous Silice | | 562 | 4.3 | 74 |
| \$D1913 | Amorphous Silica | | 395 | 6.2 | 5 |
| SD1913(c) | Amorphous Silica | | 368 | 6.5 | 42 |
| SD1868 | H-Y Zeolite | 5.3:1 | 581 | 4.4 | 27 |
| \$D2209 | Y-Zeolite dealumntd. | 33:1 | 733 | 5.5 | 99 |
| SD1867 | 4A-Zeolite | 2.0:1 | 20 | 1.5 | 0 |
| SD2006 | H-Y Zeolite | 5.2:1 | 600 | 3.5 | 33 |
| SD2210 | Hydrotalcite | | 178 | 9.3 | 10 |
| Celite 545 | Diatomaceous Earth | | <10 | 57 | 0 |

The materials labelled SD are inorganic materials available from Crosfield Ltd., Warrington, England. The suffix (c) denotes the material has been heat treated for 2hours at 700°C. Cellte 545, which is referred to in US-A-4552591, is a commercially available product available from World Minerals, Cellte UK Limited of Livingstone Road, Hessle, Hull, North Humberside, HU13 OEG. It can be seen that Cellte 545 has no affinity for the blocide and is equally as ineffective at retaining the blocide as 4A zeolite and hydrotelcite and many others which are not reported in Table 1. In contrast, the H-Y zeolite, the dealuminated form of Y-zeolite and the heat treated amorphous silicas SD1866(c), SD1913(c) all have a significant propensity to adsorb and retain OIT.

Example 2

Those carrier particles which, from Example 1, were found to exhibit the highest retention of OIT were also investigated for their capability to adsorb CIT/MIT, a higher potency blocide than OIT, and were compared with Cellte 545 and 4A-zeolite. To obtain information on the micropore system present in the materials investigated, nitrogen adsorption isotherms were determined to allow calculation of the pore area within the micropore size range of 20 to 50 angstroms. In Table II, the micropore area is listed together with the retention values for OIT and CIT/MIT. In addition, to distinguish between the retention values of the biocides and their potencies, a Retention Factor R has been calculated for the materials investigated. The Retention Factor R is the quotient A/P of the % of active ingredient A by weight remaining in the pore system after contacting with water under the conditions prescribed and the potency P of the biocide as measured on the basis of the reference microorganism, Aureobasidium poliulens, in terms of the Minimum Inhibition Concentration (MIC), which for OIT and CIT/MIT is 36 and 5 mg/i respectively.

Table !!

| Sample No. | Pore Area | OIT | CIT/MIT | Retention | Retention |
|--------------|-----------------------|-------------|-------------|-----------|-----------|
| · | m²/g | Retention | Retention % | Factor R | Factor R |
| | 20 to 50 Angstroms | % by weight | by weight | OIT | CIT/MIT |
| SD1866(c) | 211 | 74 | 15 | 2.05 | 3 |
| SD1913(c) | 48 | 42 | 15 | 1.2 | 3 |
| SD1868 | 38 | 27 | 0 | 0.75 | 0 |
| SD 1867 | 0 | 0 | 0 | 0 | 0 |
| (4A-zeolite) | | | | | |
| SD2209 | 64 | 99 | 15 | 2.75 | 3. |
| SD2206 | 48 | 33 | 0 | 0.92 | 0 |
| Cellte 545 | 0 | 0 | 0 | 0 | 0 |

It can be seen from Table II that those materials identified for their good retention of blocide in Example 1 have a Retention Factor R in excess of 0.6 for the two blocides, compared with the prior art materials Cellte 545 and 4A-zeolite for which the value is zero. The retentive materials were found to have pore areas in the pore size range 20 to 50 Angstrom of greater than 35 m²/g indicating their potential for controlled release of blocide into a substrate, such as a paint or lacquer system.

Example 3

To produce larger samples of the blockde impregnated inorganic carriers for testing in paint formulations the following method was utilised. The appropriate amount of blockde was added dropwise to the inorganic particulate carrier (500g) whilst it was being stirred in a Sirman SV6 Food Processor (available from Metcalf catering Equipment, Bleanau Ffestinlog, Gwyndd, Wales) so that

the finished product contained 27% by weight of the blocide. The blocide impregnated particulate carrier composition was then sealed in a tin to prevent the loss of volatile components and to allow the blend to equilibriate before mixing into a paint system. The following method was used to disperse the blocide carrier composition in the paint formulation.

The appropriate amount of the blockde carrier composition needed to yield a dry film concentration of 100ppm for CIT/MIT and 600ppm for OIT, was added to 1kg of the paint formulation retained in a suitably sized vessel to allow the dispersion to be agitated with a Cowles high speed disperser. This premix was the transferred to a Silverson type mixer, fitted with a medium mixing head, and mixed until a finer dispersion was obtained. Control formulations containing only (I) the free blockde and (II) free blockde plus unloaded inorganic carrier particles were prepared in exactly the same way so as to eliminate the method of dispersion as a variable. For each preparation, during dispersion on the Silverson Mixer, the temperature of the vessel was controlled between 40 to 50°C. The two model paint fomulations, one a water-borne acrylic and the other a solvent-based alkyd used in this Example had the following compositions:

Solvent-based alkyd

| | % By Weight |
|--------------------------------------|-------------|
| White Spirit | 19.5 |
| Urethane Alkyd (55% in white spirit) | 78.2 |
| Cobalt Drier | 0.4 |
| Zirconium Orier | 0.4 |
| Calcium Drier | 8.0 |
| Methyl Ethyl Ketoxime | 0.1 |
| Defoamer | 0.0 |

(Urethane Alkyd as supplied under the trade name Unithane 655W by Cray Valley of Waterloo, Machen, Newport, Gwent NP1 8YN, UK)

Water-borne acrylic

Acrylic emulsion

Dispelair CF269 added as required to counteract foaming

(Acrylic emulsion as supplied, devoid of protective blocidal additive, under the trade name Revacryl 1A by Haripw Chemical Co. and Dispetair CF269 defoamer supplied by Blackburn Chemicals)

The blocidal efficacy in the above paint formulations of the range of carriers loaded with OIT or CIT/MIT were compared with the blocides in non-loaded systems. A rapid screening test using Cladosporlum cladosporloides was used to determine the zones of inhibition around cured, painted rubber discs placed, coated face down, on a solid agar surface seeded with the fungus as described hereinbefore. Tables III and IV summarise the data obtained on the water-borne acrylic and the alkyd paint systems containing free OIT, added only as the blocide to the paint formulation(Free), OIT

. . . .

loaded onto the carrier (SDno.(L)) and free OIT together with the unloaded inorgenic carrier (Free+SDno.). SDno denotes the SD number of the inorganic carrier as listed in Tables III and IV. To highlight distinguishing features in performance, the concentration of OIT in the paint film ranged from 600 to 12000ppm for both the acrylic and the alkyd paint system, irrespective of the method of addition of the biocide to the formulation.

Table III-Acrylic Paint Formulation

| Paint Formuln. | OIT Concn. ppm | Addition Method | Zone Present | Zone Size mm |
|----------------|-------------------|--------------------|--------------|-----------------|
| P1 | 600 | Free | + | < 1 |
| P2 | 1000 | Free | + | < 1 |
| P3 | 2000 | Free | + | 7 |
| P4 | 4000 | Free | + | 10 |
| P5 | 8000 | Free | + | 18 |
| P6 | 12000 | Free | + | 20 |
| P7 | 600 | SD2209(L) | + | < 1 |
| P8 | 1000 | SD2209(L) | + | < 1 |
| P9 | 2000 | SD2209(L) | + | 3 |
| P10 | 4000 | SD2209(L) | + | 9 |
| P11 | 8000 | SD2209(L) | + | 13 |
| P12 | 12000 | SD2209(L) | + | 18 |
| P13 | 600 | Free + SD2209 | + | < 1 |
| P14 | 1200 | Free + SD2209 | + | < 1 |

Table IV-Alkyd Paint Formulation

| Paint | OIT Concn. | Addition | Zone Present | Zone Size |
|-------------|------------|-----------|--------------|-----------|
| Formulation | ppm | Method | | mm |
| P16 | 1200 | Free | . +1> | < 1 |
| P16 | 2000 | Free | + | 2 |
| P17 | 4000 | Free | + | 3 |
| P18 | 8000 | Free | + | 5 |
| P19 | 12000 | Free | + | 10 |
| P20 | 1200 | SD2209(L) | + | < 1 |
| P21 | 2000 | SD2209(L) | + | 1 |
| P22 | 4000 | SD2209(L) | + | 3 |
| P23 | 8000 | SD2209(L) | + | 4 |
| P24 | 12000 | SD2209(L) | + | 5 |

The "zone sizes" referred to above are the summed radial distances, as previously mentioned.

In Tables III and IV, "+" Indicates that there was an observable zone of fungal inactivity around the periphery of the painted disc. It can be seen that under the test conditions the response for OIT is not significant in the concentration range 600 to 1,000ppm in either paint fomulation. To measure the levels of inhibition imparted by the blocide, concentrations in the paint film needed to be increased to values in excess of 1,200ppm and, in the range 1,200 to 12,000 ppm, it is possible to see differences produced by the different modes of introducing the blockde into the paint formulation. In both formulations there is clear evidence that incorporation of blocide to the paint system adsorbed in the pore system of the inorganic particulate carrier is slowing down the response for OIT, as can be seen by comparing the size of the zones of inhibition for the paint formulations containing high concentrations of biocide. For the alkyd system at a loading of 12000ppm of OIT there is marked reduction in the zone width, from 10 to 5mm, between the peint formulation containing free biocide and the one where the blocide has been added adsorbed in the pore system of the inorganic particiate carrier. For the water borne acrylic the difference in zone widths is not as marked (20 mm compared with 18mm for paint formulations containing 12000ppm of OIT) but in general on comparing the Inhibition zones obtained for the range for the formulations containing 2000 to 8000ppm there is sufficient evidence to support the observation that, in this paint system, the response is being slowed by incorporating the biocide on the inorganic particulate carrier.

CLAIMS

- 1. A particulate composition of matter comprising porous inorganic carrier particles having blocide adsorbed within the pore system thereof and having a retention factor (as defined herein) of at least 0.6.
- A composition as claimed in Claim 1 in which the retention factor is at least 0.8.
- 3. A composition as claimed in Claim 1 or 2 in which the particles carry at least 30% of blockde solution.
- 4. A composition as claimed in any one of Claims 1 to 3 in which the particles have an activated micropore system.
- 5. A composition as claimed in any one of Claims 1 to 4 in which the particles have a pore area of at least 25 m²/g in the pore size range of from about 20 to about 50 Angstroms.
- 6. A composition as claimed in any one of Claims 1 to 5 in which the particles have a BET surface area of at least 200 m²/g
- 7. A composition as claimed in any one of Claims 1 to 5 in which the particles have a BET surface area of at least 300 m²/g.
- 8. A composition as claimed in any one of Claims 1 to 7 in which the particles have a blocide adsorption capacity of at least 10% by weight.
- 9. A composition as claimed in any one of Claims 1 to 8 in which the particles are constituted by amorphous silicas, Y-zeolites or dealuminated Y-zeolites, or a mixture of two or more of these materials.
- 10. A surface coating formulation incorporating the particulate composition as claimed in any one of Claims 1 to 9.
- 11. A formulation as claimed in Claim 10 in the form of a paint or lacquer.
- 12. A formulation as claimed in Claim 10 in the form of a water-based or organic solvent-based paint.
- 13. A sealant formulation incorporating the particulate composition as claimed in any one of Claims 1 to 9.
- 14. A surface cleaning composition incorporating the particulate composition as claimed in any one of Claims 1 to 9.

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ABSTRACT OF THE DISCLOSURE

Particulate carrier for blocide formulations

A particulate carrier material is impregnated with a blocidal formulation and serves as a vehicle for introduction of the blocide into a liquid-based media, such as a surface coating or surface cleaning compositions, in order to allow controlled release of the blocide to combat bacterial, fungal, algal or like growth for an extend period of time.